

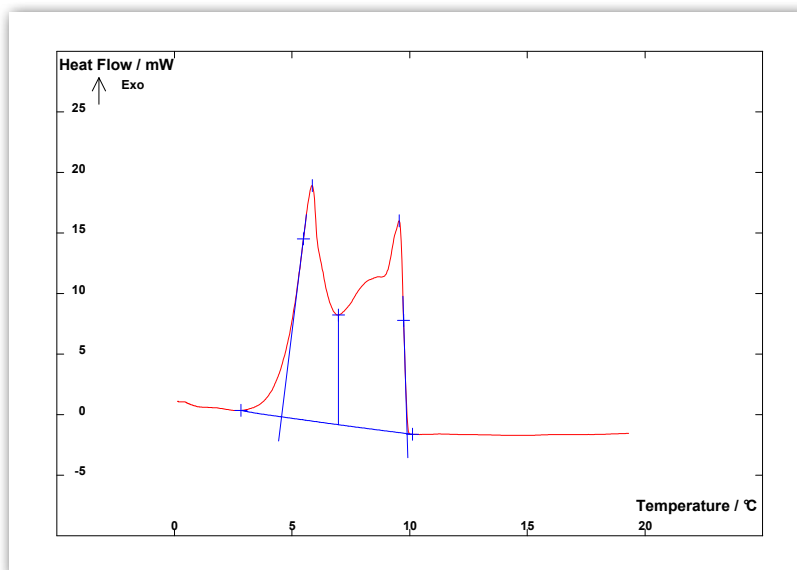
## Formation of hydrocarbon gas hydrates

### INTRODUCTION

The formation of hydrocarbon gas hydrates is a real problem for gas transportation in gas pipeline : hydrocarbon gases (methane, ethane, propane, butane..) may form hydrates, just by contact of these gases with water, at temperature ranging from 5°C to 12°C. These hydrates having the consistency of a very viscous liquid, the consequence of such a process might be a complete clogging of the pipe.

### EXPERIMENT

Sample : 8.03 g of sea water  
 High pressure flow vessel.  
 Atmosphere : mixture of methane : CH<sub>4</sub> (95 %) + propane : C<sub>3</sub>H<sub>8</sub> (5 %) under a pressure of 65 bar. The gas is bubbling inside the sea water at 2 ml/h.  
 The temperature is programmed from 25°C down to 0°C at 0.1 K/min.



### RESULTS AND CONCLUSION

The cooling curve presents different exotherms of formation of hydrocarbon hydrates with an onset temperature of 9.9°C. They are in fact two main peaks with maxima at 9.6°C and 5.9°C and respective energies of 27.6 J and 26.1 J (the two peak fractions are integrated as shown on the figure).

### INSTRUMENT

#### CALVET CRYO

**-196 °C to 200 °C**



#### HIGHEST HEAT MEASUREMENT ACCURACY

3D sensor based on thermocouples with Joule effect calibration

#### ISOTHERMAL OR TEMPERATURE SCANNING MODES

for increased flexibility and replication of real life conditions

#### CONVENIENT INTERCHANGEABLE CRUCIBLES AND CELLS

to perform even the most demanding experiments using one instrument :

- high pressure (1000bar) and high vacuum
- pressure measurement and control
- mixing/stirring experiments

#### EXTERNAL COUPLING CAPABILITY

designed to increase your research options including manometry, BET instrumentation, gas analyzers, humidity controllers and gas panels