Purity Determination by DSC

INTRODUCTION

The Van't Hoff method allows for the determination of the purity of substances based on a DSC curve. An example of this method is introduced in the present document, based on samples of Phenacetin with certied impurity levels and provided by NIST. Check the application note "ANS006 - Purity Determination by DSC – 1 (Theory)" for more details about the mathematics involved and the limitations of the method.

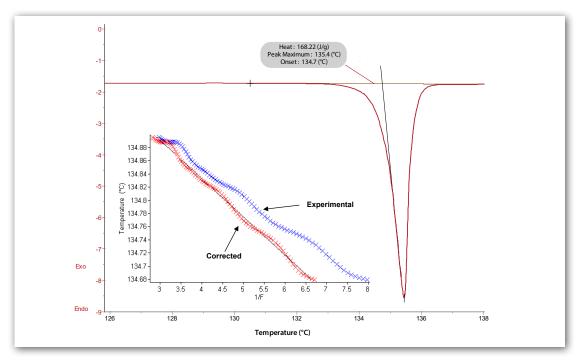


Figure 1 - Purity determination of 'pure' phenacetin by the Calisto software tool

EXPERIMENT

Sample:

Set of certied Phenacetin samples with different purity rates (pure, 99.3%, 98% and 95%)

Experimental conditions:

- Atmosphere: Nitrogen, atmospheric pressure
- Sample mass: About 2 mg in a 30 μ l aluminum crucible sealed with a lid. 110°C \rightarrow 150°C at 1°C/min

RESULTS AND CONCLUSION

The green curve on Figure 1 corresponds to the melting peak of the certied "pure" Phenacetin sample. The inset plot represents the experimental function $T_s = f(1/F)$ and the corrected function by the SONDACK method.

To = 135.07°C Correction K = 9.31 mJ or 2.78 % Slope = -0.06°C Molar purity = 99.87%

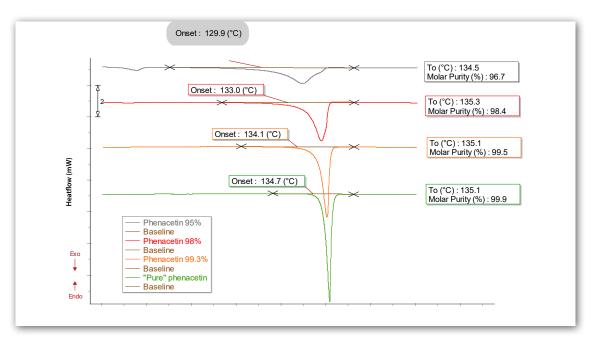


Figure 2 - Melting peaks superposition of phenacetin samples at different purity ratios

RESULTS AND CONCLUSION

Figure 2 corresponds to the melting peaks of all the certified Phenacetin samples.

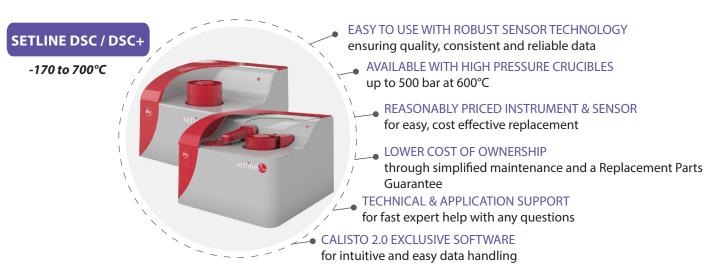
The purity determination function of Calisto Data Processing software automatically applies the Van't Hoff relationship and the SONDACK correction method, leading to the evaluation of the molar impurity percentage of the tested Phenacetin samples. Check the application note "ANS006- Purity Determination by DSC – 1 (Theory)" for more details.

When the impurity rate increases, the melting peak is spread over a larger range of temperature.

On the thermogram corresponding to phenacetin with 5% impurity, the eutectic melting peak is observable prior to the melting of phenacetin. In this case, the purity measurement is much less precise.

This observation, and the comparison of the certied purity values vs. the values obtained by DSC, can confirm that this method is accurate for relatively pure compounds (less than 2% of impurity).

INSTRUMENT



KEP